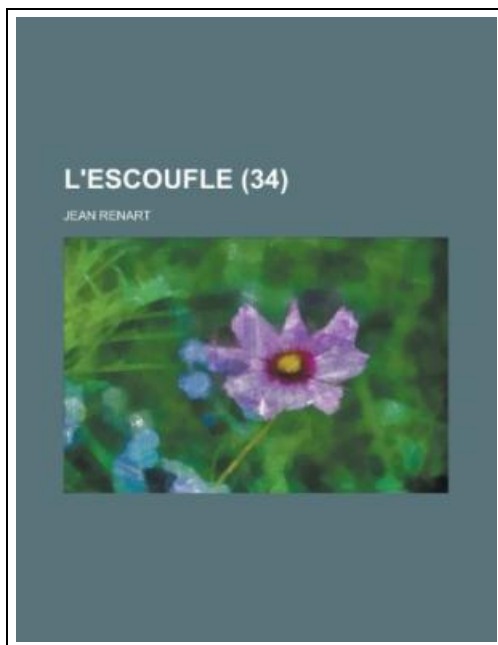


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RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 42 pages. Original publisher: Gaithersburg, MD: U. S. Dept. of Commerce, Technology Administration, National Institute of Standards and Technology, Building and Fire Research Laboratory, 1999 OCLC Number: (OCoLC)713570073 Subject: Combustion -- Measurement. Excerpt: . . . researchers which were obtained at higher hydrogen mole fractions. The present data and those of earlier investigations were numerically modeled using a mechanism based on Allen et al. 7 with a CO OH rate from Yu et al. 40 , and the CO N O direct reaction rate of Milks 2 and Matula 5 . Modeling of the flames requires the use of the direct reaction, and the present 9. 2 3 results imply a rate of 10 cm mol s at 1800 K, which corresponds to a 10 decrease in the pre-exponential factor of the Milks and Matula rate. Experiments with nitrogen dilution and over a range of suggest an activation energy near 71 kJ mole. For the moist flames, the CO OH rate also has a strong effect on the predicted burning velocity, and the rate of Yu et al. 40 provides good agreement with our data. Iron pentacarbonyl, which is the most effective flame inhibitor identified for hydrocarbon-air flames, is not effective in N O flames; in fact, 213 ppm of Fe ( CO ) actually 2 5 increases the burning velocity of the dry flames by about 25 . The promotion is believed to be due to the iron-catalyzed gas-phase reaction of N O with CO, via N O M N MO and 2 2 2 CO MO CO M, where M is Fe, FeO, or FeOH. The rate expression of Plane and 2 Rollason 49 for the former reaction with M Fe, together with estimates of the...

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