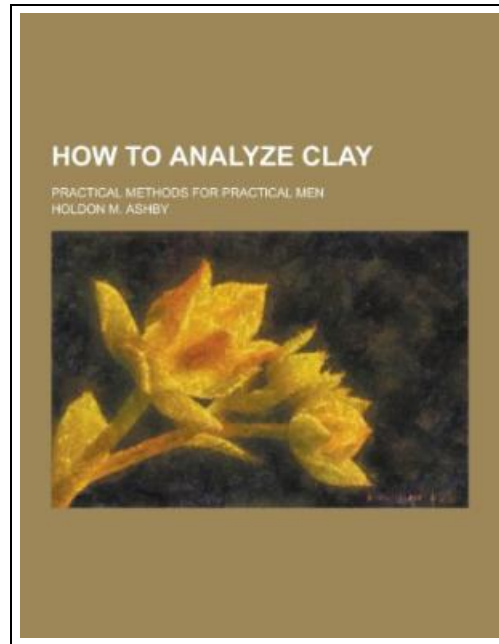


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Theclassics.us, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1897 edition. Excerpt: .and the carbon-di-oxide driven off, this may be detected by adding a few drops of water to the precipitate after it has been weighed, and testing the solution with a slip of turmeric paper. If the paper does not turn brown, everything-is vight; if it does caustic lime is present and the paper must be rinsed off into the crucible, a small lump of pure ammonium ca-bonate added, and the precipitate evaporated to dryness by placing the crucible on an asbestos mat and applying a very gentle heat. The heat must not be sufficient to cause any sputtering of the precipitate. When dry the latter should be heated on the clay triangle as before, only more carefully. After 15 minutes of a very faint red heat it may be cooled and weighed again. The ammonia carbonate is added for the purpose of supplying the lost carbonate acid gas, the ammonia and the unused ammonium carbonate being driven off at a very low temperature. Each gram of calcium carbonate contains 0.56 gr. calcium oxide, so the result must be multiplied by 0.56 to give the calcium oxide. SECOND METHOD. The precipitate is placed in the weighed crucible and the filter paper burnt, precisely as in the first method. The crucible containing the precipitate and ash is placed on the support of the blast lamp and the temperature slowly raised to a very bright glowing heat. This should be kept up for about 10 or 15...



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